

## B15 Organisms and their environment

### B15.3 Carbon Cycle

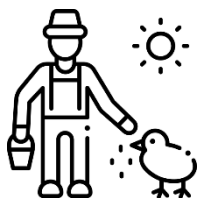
#### The Carbon Cycle

The carbon cycle keeps the amount of carbon in the atmosphere roughly balanced, but human activities (burning fuels, deforestation) are increasing CO<sub>2</sub> levels, contributing to climate change.

#### Key Processes

**Photosynthesis** Plants take in carbon dioxide (CO<sub>2</sub>) from the air. They use it to make glucose (C<sub>6</sub>H<sub>12</sub>O<sub>6</sub>) during photosynthesis.

- Equation:  $6\text{CO}_2 + 6\text{H}_2\text{O} \rightarrow \text{C}_6\text{H}_{12}\text{O}_6 + 6\text{O}_2$
- Carbon becomes part of plant biomass (e.g. leaves, stems).

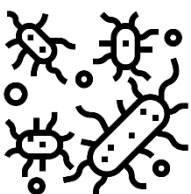


#### Feeding

- Animals eat plants and obtain carbon compounds. Carbon moves through the food chain as animals eat other animals.

#### Respiration (occurs in cells)

- Plants and animals release CO<sub>2</sub> back into the air when they respire.
- Equation:  $\text{C}_6\text{H}_{12}\text{O}_6 + 6\text{O}_2 \rightarrow 6\text{CO}_2 + 6\text{H}_2\text{O}$

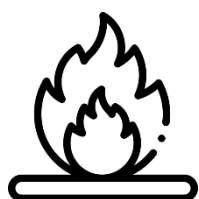


#### Decomposition

- Microorganisms (bacteria and fungi) break down dead organisms and waste.
- During decay, CO<sub>2</sub> is released through respiration of decomposers.

#### Formation of Fossil Fuels

- Some dead plants and animals are buried and compressed over millions of years.
- This forms coal, oil, and natural gas, which store carbon underground.



#### Combustion (Burning)

- Burning fossil fuels or wood releases CO<sub>2</sub> into the atmosphere.
- Equation (example):  $\text{CH}_4 + 2\text{O}_2 \rightarrow \text{CO}_2 + 2\text{H}_2\text{O}$